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Reactions. Base.

Chemical Equation.

Chemical Reaction.

Stoichiometric

Coefficients

(coefficien.... A

substance that

produces hydroxide

(OH-) ions when

dissolved.... Chemical

symbols of reactants

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and products
representing a chem....
Process by which
substances are
changed into other
substances.

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reactions. Combustion.
Neutralization. Solvent.
Decomposition. a
chemical reaction that
occurs between a fuel

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and an oxidizin.... a
Chemical reaction in
which an acid and a
base react quantita....
able to dissolve other
substances. breakdown
is the separation of a
single chemical
compound into....

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Multiple choice test:
Chemical Formulas and
EquationsTN SPI

807.9.8 - Interpret the

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results of an investigation to determine whether a physical or chemical change has

occurred. TN SPI 807.9.11 - Recognize that in a chemical reaction the mass of the reactants is equal to the mass of the products (Law of Conservation of Mass) TN SPI 807.9.10 Identify the reactants and products of a chemical reaction.

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Chemical Reactions
CHAPTER 10 What
You'll Learn You will
write chemical
equations to describe
chem-ical reactions.
You will classify and
identify chemical
reactions. You will
write ionic equa-tions
for reactions that occur
in aqueous solutions.

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Why It's Important
Chemical reactions affect you every second of every day. For example, life-sustaining chemical reactions occur con-

Chapter 10: Chemical Reactions

Chapter 8 - Chemical Reactions Notes

Chemical Reactions:

Chemical reactions are processes in which the atoms of one or more substances are

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rearranged to form different chemical compounds. How to tell if a chemical reaction has occurred (recap):
Temperature changes that can't be accounted for. o
Exothermic reactions give off energy (as in fire).

Chapter 10 - Chemical Reactions Notes

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end of the chapter. At this stage MCQs are important for the revision of concepts and missing terms related to the chapters. If you read 5 times the same chapter from NCERT, every time you will get something new ...

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Chemical reactions have the same number of atoms before the reaction as after the reaction. Balancing chemical equations is a basic skill in chemistry and testing yourself helps retain important information.

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