

# 161 Properties Of Solutions Key

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### **161 Properties Of Solutions Key**

-saturated solution -solubility -unsaturated solution -miscible  
-immiscible -supersaturated solution -henry's law

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Key Takeaways Key Points. There are two conceptual steps to form a solution, each corresponding to one of the two opposing forces that dictate solubility. The first conceptual step is dissolution, which corresponds to the force of the solvent  
-solvent and solute -solute intermolecular attractions that needs

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to be broken down.

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13.E: Properties of Solutions (Exercises) These are homework exercises to accompany the Textmap created for "Chemistry: The Central Science" by Brown et al. 13.S: Properties of Solutions (Summary) A summary of the key concepts in this chapter of the Textmap created for "Chemistry: The Central Science" by Brown et al.

## **13: Properties of Solutions - Chemistry LibreTexts**

Title: PowerPoint Presentation Author: Debbie Munson Created Date: 4/29/2014 8:03:20 AM

## **Chapter 16**

Section 16.1 Properties of Solutions 471 16.1 Properties of Solutions Guide for Reading Key Concepts • What factors determine the rate at which a substance dissolves? • How is

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solubility usually expressed? • What conditions determine the amount of solute that will dissolve in a given solvent? Vocabulary saturated solution solubility

### **16.1 Properties of Solutions 16**

Properties of a solution • A solution is a homogeneous mixture. • The particles of a solution are smaller than 1 nm ( $10^{-9}$  metre) in diameter. So, they cannot be seen by naked eyes. • Because of very small particle size, they do not scatter a beam of light passing through the solution. So, the path of light is not visible in a solution.

### **What is a solution? Write its properties.**

Section 16.1 Properties of Solutions 471 16.1 Properties of Solutions Guide for Reading Key Concepts What factors determine the rate at which a substance dissolves? How is solubility usually expressed? What conditions determine the

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amount of solute that will dissolve in a given solvent? Vocabulary saturated solution solubility

### 16.1 Properties of Solutions 16

Chapter 16 Solutions 167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471-477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471-472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2. 1.

### 05 Chem GRSW Ch16.SE/TE

24) A solution containing 10.0 g of an unknown liquid and 90.0 g water has a freezing point of  $-3.33^{\circ}\text{C}$ . Given  $K_f = 1.86^{\circ}\text{C}/m$  for water, the molar mass of the unknown liquid is \_\_\_\_\_ g/mol.

A)161 B)69.0 C)62.1 D)333 E)619 25) A solution is prepared by dissolving 0.60 g of nicotine (a nonelectrolyte) in water to make

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12 mL of solution.

### **A.P. Chemistry Practice Test: Ch. 11, Solutions MULTIPLE**

...

Chemistry: 16.1 Properties Of Solutions - Review 12 Questions |  
By Gorantes | Last updated: Jan 25, 2013 | Total Attempts: 97  
Questions All questions 5 questions 6 questions 7 questions 8  
questions 9 questions 10 questions 11 questions 12 questions

### **Chemistry: 16.1 Properties Of Solutions - Review ...**

Title: Microsoft Word - Chapter 15 SR Key.docx Author: Krista  
Munoz Created Date: 5/5/2013 12:18:36 PM

### **Chapter 15 SR Key**

Key Concept Check breaks the bonds between the elements. 6.  
Identify In what three ways do compounds differ from mixtures?  
5. Apply Seawater is a mixture of mainly salts and water. What



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properties of these substances can you observe in seawater?  
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